

Development of Next-generation Wide-bandgap Semiconductors

Kazuyuki Hirama, Yoshitaka Taniyasu, Hideki Yamamoto, and Kazuhide Kumakura

Abstract

Cubic boron nitride (c-BN) is a wide-bandgap semiconductor with the highest breakdown field among semiconductors. It has the potential to dramatically improve efficiency in power devices. Although c-BN is a metastable material, we have heteroepitaxially grown high-quality c-BN thin films by developing a unique growth technique. We have also used doping to control the electrical conductivity of the c-BN films. These accomplishments are major steps toward the fabrication of c-BN-based power devices.

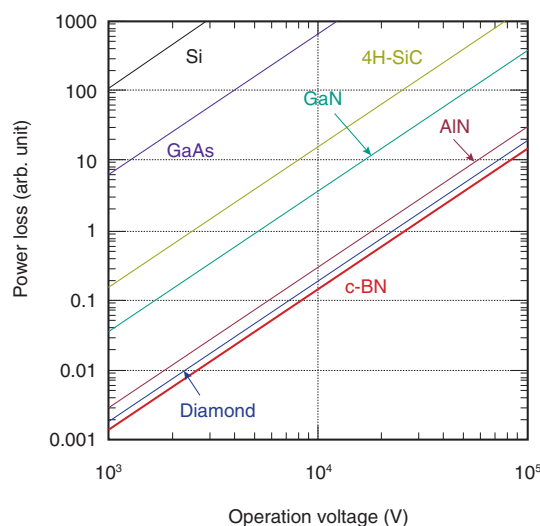
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1. Power device applications of wide-bandgap semiconductors

Wide-bandgap semiconductors are expected to serve as components of high-efficiency power devices, in which electrical power loss is significantly lower than in conventional semiconductor-based power devices. Here, silicon (Si) and gallium arsenide (GaAs) are typical conventional semiconductors. The main functions of power devices include regulation of direct current (DC) voltage, frequency conversion of alternating current (AC) voltage, and conversion between AC and DC. For example, in AC adapters used in personal computers and mobile phones, power devices convert the voltage from AC 100–200 V to DC ~20 V.

The relationship between power loss and operation voltage of power devices consisting of various semiconducting materials is shown in **Fig. 1**. Power loss is determined by the on-resistance of power devices; it becomes smaller with lower on-resistance. Increasing the carrier concentration and reducing the current-path length are both effective in reducing the on-resistance, and consequently the power loss.

Such conditions are more readily fulfilled when the breakdown field^{*1} of the constituent semiconductor is higher. The high breakdown fields of wide-bandgap semiconductors, at least one order of magnitude



AlN: aluminum nitride, c-BN: cubic boron nitride, SiC: silicon carbide, GaN: gallium nitride

Fig. 1. Relationship between power loss and operation voltage.

^{*1} Breakdown field: When the electrical field is applied to an insulator or a semiconductor, the current drastically increases above a threshold value of the electrical field (i.e., the breakdown occurs). The threshold electrical field is defined as the breakdown field. Values of the inherent breakdown field vary by material.

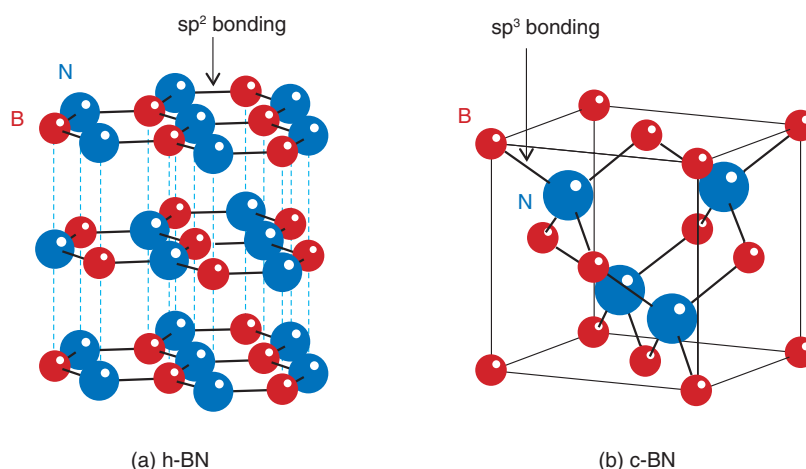


Fig. 2. Crystal structures of (a) hexagonal BN (h-BN) and (b) c-BN.

higher than those in Si or GaAs, means that the on-resistance of the power devices can be reduced by three orders of magnitude. As a result, wide-bandgap semiconductor-based power devices can operate more efficiently with lower power losses. Some wide-bandgap semiconductor-based power devices are already in practical use, as exemplified by silicon carbide (SiC)- or gallium nitride (GaN)-based ones, including high-efficiency inverters for trains.

At NTT Basic Research Laboratories, we have been studying next-generation wide-bandgap semiconductors such as cubic boron nitride (c-BN), aluminum nitride (AlN), and diamond. In the following sections, we describe the growth of c-BN thin films with the highest breakdown field among semiconductors and describe how we control their electrical conductivity. The power loss in c-BN-based power devices is expected to be four orders of magnitude lower than those in Si- or GaAs-based ones and at least one order of magnitude lower than those in SiC- or GaN-based ones (Fig. 1).

If such high-efficiency c-BN-based power devices become available, power loss can be significantly reduced in electrical vehicles, railways, and solar-power and wind-power generation systems. Accordingly, research aimed at developing c-BN-based power devices is underway, as this approach uses fewer natural resources and will contribute to constructing a sustainable society on a global scale.

2. Growth of c-BN epitaxial films using unique growth technique

Boron nitride (BN) is a compound semiconductor composed of boron and nitrogen. BN can have several crystal structures depending on the bonding character and stacking sequence. Representative crystal structures of BN are shown in **Fig. 2**. The thermodynamically stable phase under standard temperature and pressure is sp^2 -bonded hexagonal BN (h-BN). In contrast, sp^3 -bonded c-BN is a high-pressure and high-temperature stable phase, that is, a metastable phase under standard temperature and pressure.

To utilize c-BN with a high breakdown field for power devices, first of all, the growth of phase-pure (i.e., composed of only the sp^3 -bonded phase) c-BN films is necessary. Ion irradiation to the growth surface is known to be effective for forming the metastable sp^3 -bonded c-BN phase. Nevertheless, growth of a phase-pure c-BN thin film has been hampered by partial formation of the thermodynamically stable sp^2 -bonded BN phase. The sp^2 -bonded BN phase has a disordered h-BN structure and is called turbostratic BN (t-BN). The inclusion of the t-BN phase prevents epitaxial growth and renders the resultant BN films polycrystalline ones.

Despite the above-mentioned difficulties, we have grown phase-pure c-BN films by developing a unique growth technique, that is, ion-beam-assisted molecular beam epitaxy (MBE) (**Fig. 3**). Boron (B) atoms are supplied from an electron beam evaporator, while nitrogen radicals (N^*) are supplied from an RF (radio

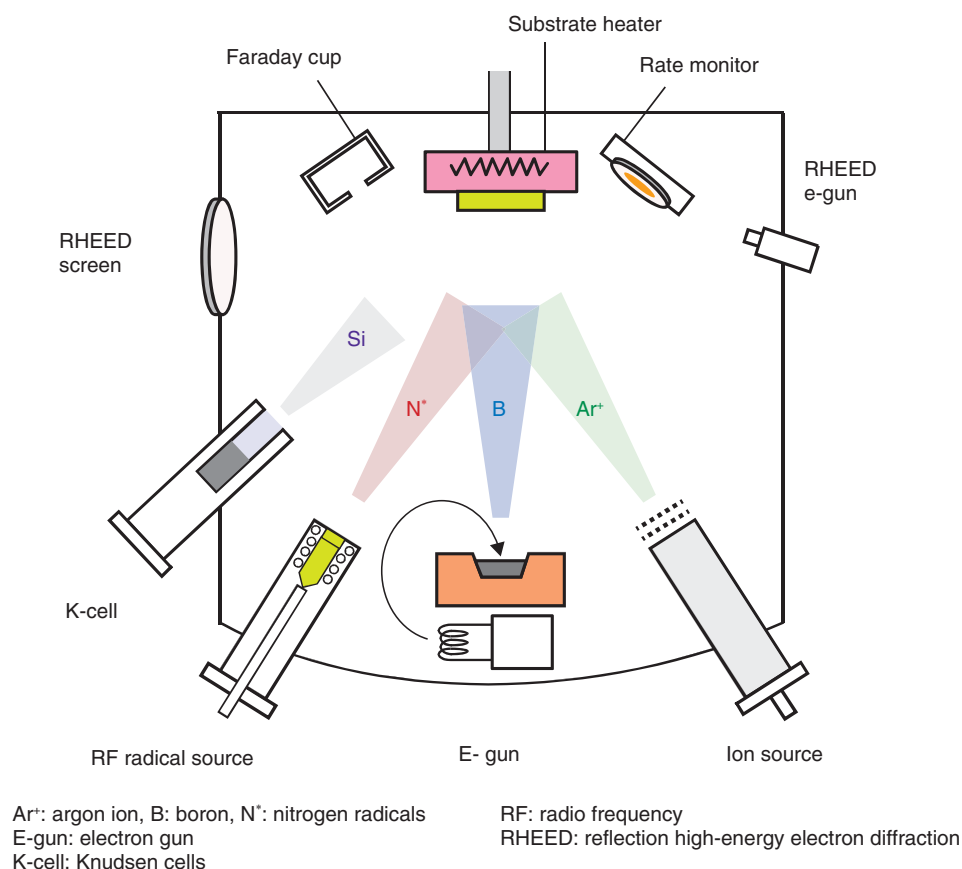


Fig. 3. Schematic of ion-beam-assisted MBE.

frequency) radical generator. At the same time, the growth surface is irradiated with argon ions (Ar⁺). The Ar⁺/B ratio was found to play a crucial role in forming the sp³-bonded BN [1, 2]. The substrate is diamond with a small lattice mismatch with c-BN.

To investigate the effect of the Ar⁺ irradiation during the BN growth on the bonding characteristics and crystal structures of the resultant BN films, we first grew BN films without and with Ar⁺ irradiation. The Fourier transform infrared spectroscopy (FT-IR) absorption measurement^{*2} of the samples prepared without and with Ar⁺ irradiation is shown in Fig. 4. In the absence of Ar⁺ irradiation, two absorption peaks were observed at around 800 cm⁻¹ and 1380 cm⁻¹, which are attributed to sp²-bonded BN. No absorption peak related to sp³-bonded BN was observed.

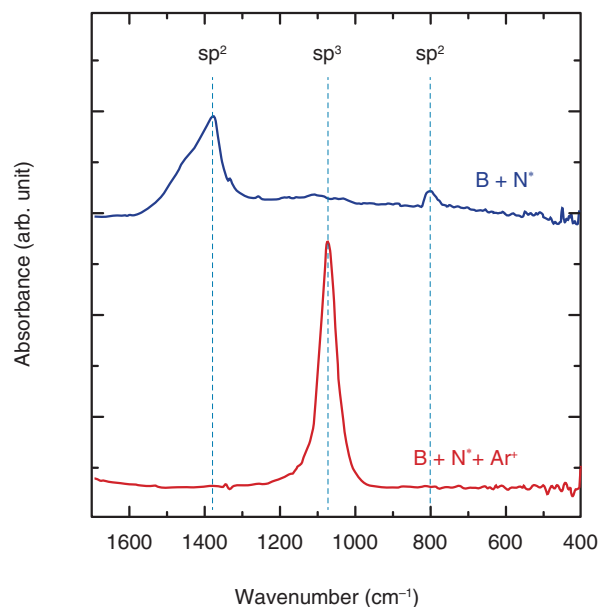


Fig. 4. FT-IR absorption spectra of BN films.

*2 FT-IR absorption measurement: An optical measurement method to characterize specific infrared (IR) absorbance in a material from the difference in intensities of transmitted and incident light. The detected signal is converted to the spectrum by Fourier transformation (FT).

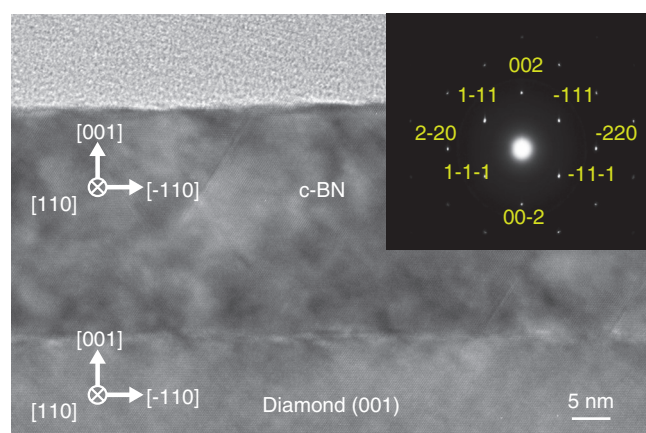


Fig. 5. Cross-sectional TEM image of c-BN film and selective-area electron diffraction pattern of c-BN region.

This means that the sp^2 -bonded BN film was grown.

In contrast, in the presence of Ar^+ irradiation, only an absorption peak was observed at around 1070 cm^{-1} , which is attributed to the sp^3 -bonded BN. No other peaks originating from sp^2 -bonded BN were observed. Therefore, the grown film is phase-pure sp^3 -bonded BN. Accordingly, the ion irradiation allows for a selective formation of the sp^3 -bonded BN. We surmise that the kinetic energy and/or momentum transferred from the Ar^+ ions to the growth surface may contribute to the preferential formation of the sp^3 -bonds in BN.

To identify the crystal structure, we performed cross-sectional TEM (transmission electron microscopy) measurement for the BN film prepared with the Ar^+ irradiation, which was phase-pure (sp^3 -bonded BN) judging from the FT-IR measurement (Fig. 5). The BN film is grown uniformly from the heterointerface (interface between the diamond substrate and the BN film) to the surface. The inset in Fig. 5 shows a selective-area electron diffraction pattern of the BN film. All the diffraction spots are assigned to those for single-crystal c-BN (001) taken along the [110] zone axis. No other spots or ring patterns attributed to sp^2 -bonded BN are observed. Therefore, we can conclude that the single-crystal c-BN (001) film is heteroepitaxially grown on the diamond (001) substrate [3].

3. Electrical conductivity control in c-BN films

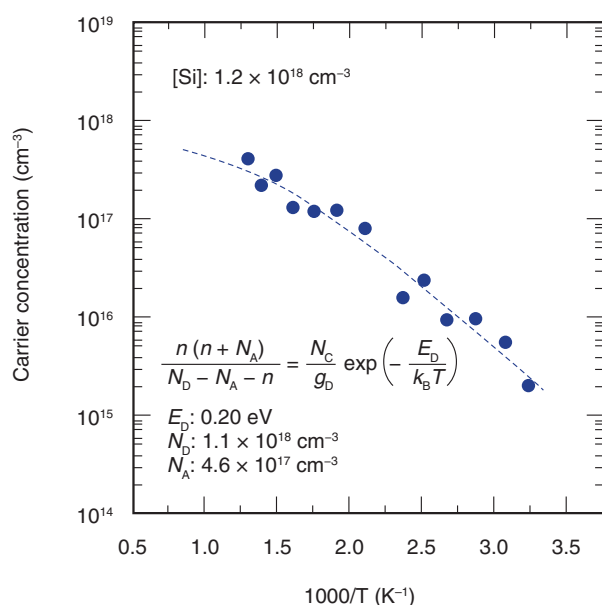
To fabricate power devices, it is necessary to grow n-type and p-type c-BN films and control their electrical conductivity. For c-BN, Si and S (sulfur) serve as donors, while Be (beryllium) serves as an acceptor.

We carried out the n-type doping first; Si was supplied as a dopant during the epitaxial growth of the c-BN (001) films by using a K-cell (Knudsen cell).^{*3}

The dependence of carrier concentration on temperature for a Si-doped c-BN film with a Si concentration ($[Si]$) of $1.2 \times 10^{18}\text{ cm}^{-3}$ is shown in Fig. 6. The carrier concentration monotonically increases with the measurement temperature. The carrier concentrations at room temperature (RT) and 500°C are respectively $2.1 \times 10^{15}\text{ cm}^{-3}$ and $6.0 \times 10^{17}\text{ cm}^{-3}$. From the fitting with the equation for the charge neutrality condition (Fig. 6, inset), the donor ionization energy (E_D), donor concentration (N_D), and acceptor concentration (N_A) are respectively estimated to be 0.20 eV, $1.1 \times 10^{18}\text{ cm}^{-3}$, and $4.6 \times 10^{17}\text{ cm}^{-3}$. Because N_D is almost the same as $[Si]$, most of the doped Si atoms are considered to be substitutionally incorporated into boron sites, where they then serve as donors. Remarkably, E_D of 0.20 eV is lower than those for other next-generation wide-bandgap semiconductors such as AlN and diamond. The low E_D in c-BN is promising for higher carrier concentration and lower resistivity, suggesting the potential of c-BN to exceed AlN or diamond, once a c-BN-based power device is achieved.

The dependence of the resistivity on $[Si]$ at RT for the Si-doped c-BN films is shown in Fig. 7. For comparison, the resistivity of the non-doped c-BN film is also plotted in Fig. 7. The resistivity of the non-doped c-BN film is as high as $\sim 10^8\text{ }\Omega\cdot\text{cm}$, while the Si-doped

^{*3} Knudsen cell: An evaporation cell, which can supply a molecular beam of a material at a precisely controlled evaporation rate by tuning the cell temperature.



E_D : donor ionization energy
 N_D : donor concentration
 N_A : acceptor concentration

Fig. 6. Temperature dependence of carrier concentration in Si-doped c-BN film.

c-BN film with a [Si] of $1.5 \times 10^{19} \text{ cm}^{-3}$ shows low resistivity of $\sim 260 \Omega\text{-cm}$. The resistivity decreases as [Si] increases, demonstrating the successful control of electrical conductivity by varying the dopant concentration, in this case, [Si].

4. Summary

We described the recent progress in research on c-BN at NTT Basic Research Laboratories. We have achieved the epitaxial growth of single-crystal c-BN thin films and control of electrical conductivity in n-type Si-doped c-BN films. These accomplishments

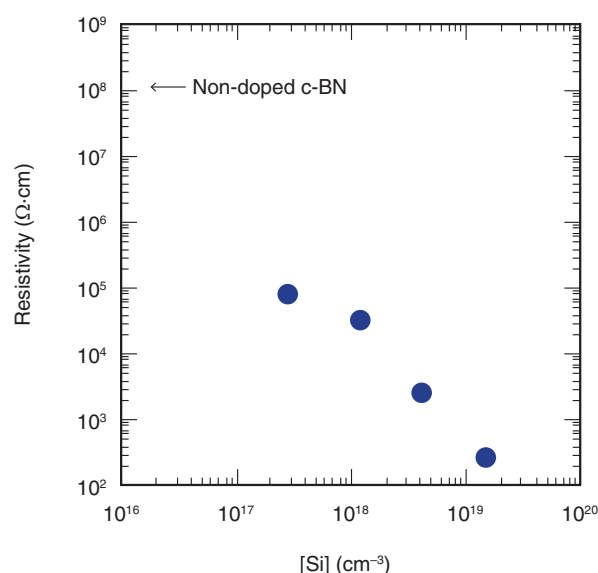


Fig. 7. Dependence of resistivity on Si concentration for Si-doped and non-doped c-BN films.

are important steps toward the fabrication of electron devices. Further improving crystalline quality and p-type doping will pave the way to achieving high-performance c-BN-based power devices.

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**Kazuyuki Hirama**

Senior Research Scientist, Thin-Film Materials Research Group, NTT Basic Research Laboratories.

He received a B.S., M.S., and Ph.D. in electrical engineering from Waseda University, Tokyo, in 2004, 2006, and 2008. He joined NTT Basic Research Laboratories in 2009. His current research interests are the growth of wide-bandgap semiconductors (c-BN, AlN, and diamond) and their device applications. He received the award for encouragement of research at IUMRS-ICA (International Union of Materials Research Societies International Conference in Asia) 2014, outstanding presentation and best poster awards at Diamond Symposium 2010 and 2012, and the young scientist presentation award at the 2008 Japan Society of Applied Physics (JSAP) meeting. Dr. Hirama is a member of JSAP.

**Yoshitaka Taniyasu**

Senior Researcher, Group Leader of Low-Dimensional Nanomaterials Research Group, NTT Basic Research Laboratories.

He received a B.E., M.S., and Dr. Eng. in electrical engineering from Chiba University in 1996, 1998, and 2001. He joined NTT Basic Research Laboratories in 2001. He has been engaged in wide-bandgap semiconductor research. He received the Young Scientist Award at the 2007 Semiconducting and Insulating Materials Conference, the Young Scientist Award of the International Symposium on Compound Semiconductors 2011, the Commendation for Science and Technology by the Minister of Education, Culture, Sports, Science and Technology of Japan (the Young Scientists' Prize) in 2011, and the Japan Society for the Promotion of Science Prize in 2019.

**Hideki Yamamoto**

Senior Research Scientist, Supervisor, Executive Manager of Materials Science Laboratory, NTT Basic Research Laboratories.

He received a B.S., M.S., and Ph.D. in chemistry from the University of Tokyo in 1990, 1992, and 1995. He joined NTT in 1995. His principal research fields are thin film growth, surface science, and superconductivity. He was a visiting scholar at the Geballe Laboratory for Advanced Materials, Stanford University, USA (2004–2005). He received the 2nd Young Scientist Presentation Award (1997) by JSAP and the 20th Superconductivity Science and Technology Award (2017) from the Forum of Superconductivity Science and Technology, the Society of Non-traditional Technology. He is a member of JSAP, the Physical Society of Japan, the Japan Society of Vacuum and Surface Science, the American Physical Society, and the Materials Research Society.

**Kazuhide Kumakura**

Group Leader of Thin-Film Materials Research Group, Materials Science Laboratory, NTT Basic Research Laboratories.

He received a B.E., M.E., and Ph.D. in electrical engineering from Hokkaido University in 1993, 1995, and 1998. He joined NTT Basic Research Laboratories in 1998. His current research interests include crystal growth of nitride-based semiconductors and their optoelectronic and magnetic device applications. He is a member of JSAP.